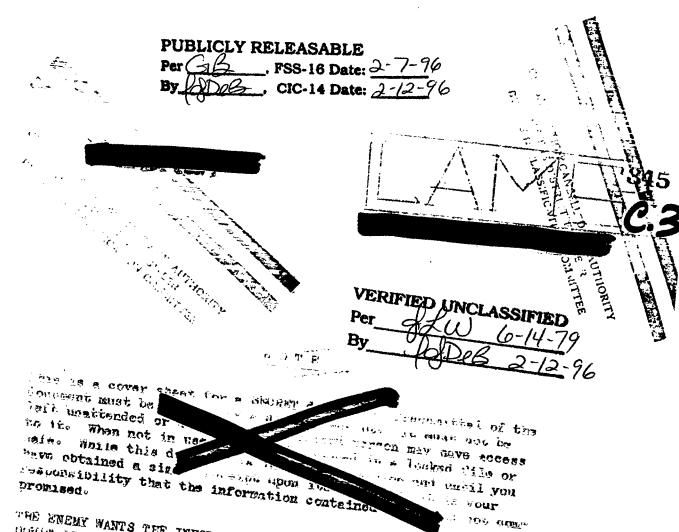
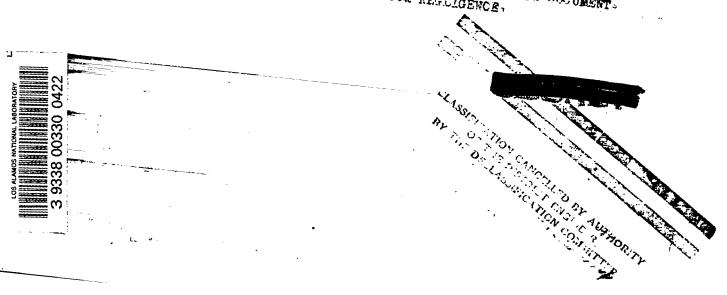
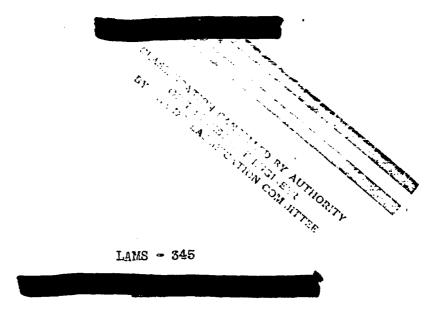
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SPIN-DEPENDENT PART OF THE NEUTRON - DEUTERIUM CROSS SECTION

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ABSTRACT

A matrix method for the spin-dependent part of the neutron- of deuterium scattering cross section is worked out in detail. The method follows closely the treatment of Condon and Shortley for the evaluation of the dot product of matrices.

SPIN-DEPENDENT PART OF THE NEUTRON - DEUTERIUM CROSS' SECTION

We wish to present a method for the spin-dependent part of the scattering cross section for slow neutrons from ortho and para deuterium. To introduce the problem let us quote from a forthcoming paper by Schwinger and Hammermesh 1:

"We wish to calculate the differential cross section for a scattering process in which a neutron with momentum p^o collides with a D₂ molecule with momentum -p^o in the internal state specified by the vibrational, rotational and spin quantum numbers v, J, S, thereby producing a neutron with momentum p which has been scattered through the angle H into the solid angle d.c., leaving the molecule in the state characterized by -p, v^o, J^o, S^o. The differential cross section, as computed by the Born approximation, is

$$\sigma_{J^{\bullet}, \mathbf{v}^{0}, \mathbf{S}^{0}; J, \mathbf{v}, \mathbf{S}} (\mathcal{G}) d\Omega = \frac{16}{25} \frac{p}{p^{0}} \frac{1}{2(25+1)(2J+1)} \frac{1}{m^{1}, m^{1}_{J}, m^{1}_{S}} \frac{1}{m_{1} m_{J}, m_{S}}$$

$$\left(\mathcal{V}_{S} \frac{MV}{2\pi N^{2}} 2 \quad U \mathcal{V}_{1} \right) \quad \frac{2}{d \Omega}$$

$$(1)$$

which differs formally for the corresponding H_2 molecule cross section $^{2/}$ only by the replacement of the numerical factor 4.9, the square of the reduced mass of the neutron - H_2 molecule system, by 16/25, the square of the reduced mass of the neutron - D_2 molecule system. These reduced mass factors arise

^{2).} Schwinger and Teller, Phys Rev 52, 1937, 286-295 equation 21



^{1).} Schwinger and Hammermesh, the scattering of slow neutrons by ortho and deuterium. To be published in the Physical Review.

-4-

in calculating the number of final neutron states per unit range of the total energy, and in the value of the neutron flux relative to the molecule.

Now we know that the interaction can be written as 3

$$U = -\frac{2\pi \hbar^2}{M} \left[\left\{ 4a_{3/2} + 2a_{1/2} \right\} + 2(a_{3/2} - a_{1/2})\sigma_n^{\circ}(\sigma_1 + \sigma_2) \right] \left\{ \delta(\mathbf{r}_n - \mathbf{r}_1) + \delta(\mathbf{r}_n - \mathbf{r}_2) \right] - \frac{2\pi \hbar^2}{M} \left[2(a_{3/2} - a_{1/2})\sigma_n^{\circ}(\sigma_1 - \sigma_2) \right] \left[(\mathbf{r}_n - \mathbf{r}_1) - (\mathbf{r}_n - \mathbf{r}_2) \right]$$
(2)

where M is the mass of the neutron; $a_{3/2}$ and $a_{1/2}$ are the scattering amplitudes for a neutron - deuteron quadruplet interaction and a neutron - deuteron doublet interaction respectively; σ_{1} and σ_{2} are the spin vectors of the deuterons; σ_{n} is the spin vector of the neutron; r_{1} , r_{2} and r_{n} are the respective position vectors and is the Dirac delta function. Units are such that the spin of the neutron is one half. The nature and use of these vectors is exactly as that given in Condon and Shortley "Theory of Atomic Spectra", Cambridge 1935 (hereafter referred to as C and S) and we follow their method in deriving the interaction matrices.

$$\begin{array}{c}
a = \frac{4a_{3}/2 + 2a_{1}/2}{B = 2(a_{3}/2 - a_{1}/2)}
\end{array}$$
(3)

thus $U = -\frac{2\pi\hbar^2}{M} \left[\alpha + \beta \sigma_n^{\circ} (\sigma_1 + \sigma_2) \right] \left[(r_n - r_1) + (r_n - r_2) \right]$ $= \frac{2\pi\hbar^2}{M} \left[\beta \sigma_n^{\circ} (\sigma_1 + \sigma_2) \right] \left[\delta (r_n - r_1) - \delta (r_n - r_2) \right]$ (4)

^{3).} One way of deriving this formula is given by Schwinger and Hammermesh and leads to their equation (6). The apparent difference is due to their definition of σ , in particular $\sigma_n = (1/2)\sigma_{\text{Schwinger}}$



We note that U is divided into a symmetrical and anti-symmetrical part and rewrite U as

Further we may rewrite the summation part of (1) using this symbolism as

$$\frac{\sum_{\mathbf{m}^{1},\mathbf{m}_{J}^{1},\mathbf{m}_{S}^{1}} \sum_{\mathbf{m}_{1}\mathbf{m}_{J},\mathbf{m}_{S}^{1}} \left| \mathbf{y}_{S}^{2} \frac{\mathbf{m}^{V}}{2\pi K^{2}} \right|^{2} = A_{1}B_{1} + A_{2}B_{2}$$
 (6)

where

$$A_{1} = \frac{\sum_{m^{1}, m_{S}^{1}} \sum_{m_{p}m_{S}} \left| (m_{1}m_{S} \ U_{1} \ m_{p}^{1} \ m_{S}^{1}) \right|^{2}}{A_{2} = ssme \text{ with } \left| U_{2} \right|}$$
(7)

and

$$B_{1} = \sum_{m_{J}} \sum_{m_{\gamma 0}} \left| \int_{\xi} \delta(z) \right|_{1} d\tau$$

$$B_{2} = \text{same with } \delta(z)$$
(8)

where and are the final and initial spacial wave functions of the whole system and is their coordinate space.

In the following we wish to show the method for evaluating ${\rm A_1}$ and ${\rm A_2}\circ$ let us recall that

m = z component of neutron spin

 $m_{\vec{d}}$ = z component of the rotation vector of the molecule

 $m_S = z$ component of $\sigma_1 + \sigma_2$

Our problem is essentially to obtain the elements of the matrices $\sigma_{n} \circ (\sigma_{14}\sigma_{2})$

$$J_{1} = \sigma_{1} + \sigma_{2}$$

$$J_{2} = \sigma_{n}$$

$$J_{2} = \sigma_{1} + \sigma_{2} + \sigma_{n}$$

$$J_{2} = \text{the z component of } J$$

and small letters denote the eigenvalues of these variables. We note that our J_2 and σ_1 now satisfy the conditions for vectors P and Q as given by C and S

page 70 where a scheme for the evaluation of P-Q is giveno

= the remaining variables completing the set

The matrix elements of P°Q are seen to be evaluated in terms of two quantities of the form (--; Q; --) and (--; P; --). The evaluation of these quantities in detail is given by equation C and S 10^3 2a and 10^3 2b°. Their substitution into the evaluation of 12^3 2 then gives P°Q°.

For a given molecular spin S, the values of J, J_1 and J_2 will determine the role of the second deuteron spin so that we may apply the argument of C and S regarding relative phases (cf C and S, last paragraph of page 69) with the following results: for elements diagonal in j, j_1 and j_2 the value of $\sigma_n^{\circ}\sigma_1$ and $\sigma_{n^{\circ}\sigma_2}$ are the same; elements off the diagonal are the negative of each other. Therefore the diagonal elements of $\sigma_{n^{\circ}}(\sigma_1^{\circ}\sigma_2)$ are zero and the off-diagonal elements of $\sigma_{n^{\circ}}(\sigma_1^{\circ}\sigma_2)$ are zero.

This means that collisions in which the total molecular spin changes are due only to interaction terms of the form $\sigma_n \circ (\sigma_1 - \sigma_2)$ i.e. only due to U_2 , the

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antisymetrical part of the interaction. Thus for the case in which total molecular spin changes $A_1 = 0$ and we need to evaluate only $A_2 = 0$ and we need to evaluate only $A_1 = 0$ and we need to evaluate only $A_1 = 0$ and $A_2 = 0$ and $A_3 = 0$ and $A_4 = 0$ an

The possible values of S for deuterium are 0, 1 and 2, Ret us now treat the individual cases by the method outlined above.

I. SYLMETRIC

$$A_{2} = 0 \text{ and}$$

$$A_{1} = \sum_{\mathbf{j}, \mathbf{j}_{z}} \sum_{\mathbf{j}^{1}, \mathbf{j}_{z}^{1}} \left| \left(\mathbf{j}_{0} \mathbf{j}_{z} \right) a + \beta \sigma_{\mathbf{n}^{0}} \left(\sigma_{1} + \sigma_{2} \right) \left| \mathbf{j}_{s}^{1} \mathbf{j}_{z}^{1} \right|^{2}$$
(9)

By our previous arguments regarding $\sigma_{n^{\circ}}(\sigma_{1}+\sigma_{2})$ the matrix $a+\sigma_{n^{\circ}}(\sigma_{1}+\sigma_{2})$ is diagonal (a being just a constant). Substitution into the formula for PoQ also shows that two elements with the same j and different j₂ are equal. The summation over j₂ therefore gives

$$A_{1} = \sum_{j} (2j+1) \left\{ \alpha + \beta_{\sigma_{n}} \cdot (\sigma_{1} + \sigma_{2}) \right\}^{2}$$
(10)

Case I-1:

S goes from 0->0

The only possible eigenvalues are

$$j = 1/2$$
, $j_1 = 0$, $j_2 = 1/2$

Substitution of these values in PoQ gives $\sigma_{n^{\circ}}(\sigma_{14\sigma_{2}}) = 0$

Hence

$$A_1 = 2a^2$$

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Case I - 2:

S goes from 1 -> 1

There are two possible sets of eigenvalues

$$j = 1/2, \quad j_1 = 1, \quad j_2 = 1/2$$

Here $\sigma_n \circ (\sigma_1 + \sigma_2) = 1$

or
$$j = 11/2$$
, $j_1 = 1$, $j_2 = 1/2$

Here $\sigma_{\mathbf{n}^{\circ}}(\sigma_{\mathbf{1}^{+}\sigma_{\mathbf{2}}}) = 1/2$

Hence $A_1 = 6 a^2 + 3\beta^2$

Case I - 3

S goes from 2---)2

There are two possible sets of eigenvalues

$$j = 1 \sqrt{2}, j_1 = 2, j_2 = 1/2$$

Here $\sigma_{\mathbf{n}} \circ (\sigma_{1} + \sigma_{2}) : -3/2$

or
$$j = 2 \frac{1}{2}, j_1 = 2, j_2 = \frac{1}{2}$$

Here $\sigma_{n} \circ (\sigma_{1} + \sigma_{2}) = 1$

Hence $\Lambda_1 = 10 \alpha^2 + 15 \beta^2$





II Anti-Symmetrical

By analagous arguments to the symmetrical case we obtain A

$$A_2 = \sum_{j} (2_{j+1}) \left\{ \beta_{\sigma_n} \cdot (\sigma_1 - \sigma_2) \right\}^2$$

In all anty-symmetrical cases (-P:-) has the value 1/2 and on different values as listed in each case.

Case II - 1

S goes from 0 -> 1

For this case (----) = 12/3

There are two possible sets of eigenvalues

$$j = 1/2$$
, $j_1 = 0 \rightarrow 1$, $j_2 = 1/2$

Here
$$\sigma_{n} \cdot (\sigma_1 - \sigma_2) = \sqrt{2}$$

or
$$j = 1 \frac{1}{2}, \quad j_1 = 0 \longrightarrow 1, \quad j_2 = \frac{1}{2}$$

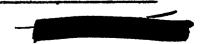
Here
$$\sigma_n \circ (\sigma_3 - \sigma_2) = 0$$

Hence
$$A_2 = 4\beta^2$$

Case II - 2

S goes from 1-70

This is analagous to the 0-1 case, and yields







S goes from 1->2

For this case
$$(-iQi-)=\frac{1}{2\sqrt{3}}$$

There are two possible sets of eigenvalues

$$j = 1 \frac{1}{2}, \quad j_1 = 1 \rightarrow 2, \quad j_2 = 1 \stackrel{1}{2}$$

Here $\sigma_{n^{\circ}}(\sigma_{1}^{-}\sigma_{2}) = \sqrt{15}$

or
$$j = 2 \frac{1}{2}, \quad j_1 = 1 \rightarrow 2, \quad j_2 = \frac{1}{2}$$

Here
$$\sigma_{\mathbf{n}} \cdot (\sigma_{\mathbf{1}} - \sigma_{\mathbf{2}}) = 0$$

Hence

The above method is interesting largely because it can be extended to other elements with relative ease.

The calculation of the hydrogen case is almost no extra work once the deuterium case is dealt with. If the interaction is written of the form of equation (4) where a and Simply have a different meaning then the values of A₁ and A₂ just evaluated change only because of a changed (—:P:—) and (—:Q:—) in the case of hydrogen.

We are contemplating the extension of this method to $\mathtt{CH_4}$ and possibly $\mathtt{NH_5}$.



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